

## higher education \& training

Department:
Higher Education and Training REPUBLIC OF SOUTH AFRICA

## T1390(E)(A9)T

NATIONAL CERTIFICATE POWER MACHINES N5 (8190035)

## 9 April 2018 (X-Paper) <br> 09:00-12:00

REQUIREMENTS: Steam tables (BOE 173)
Superheated Steam Tables (Appendix to BOE 173)
Calculators may be used.
Candidates need drawing instruments.

This question paper consists of 5 pages and a formula sheet of 3 pages.

## DEPARTMENT OF HIGHER EDUCATION AND TRAINING REPUBLIC OF SOUTH AFRICA

NATIONAL CERTIFICATE
POWER MACHINES N5
TIME: 3 HOURS
MARKS: 100

## INSTRUCTIONS AND INFORMATION

1. Answer ALL the questions.
2. Read ALL the questions carefully.
3. Number the answers according to the numbering system used in this question paper.
4. Write neatly and legibly.

## QUESTION 1

1.1 Name THREE compounds, with their chemical symbols, that are formed when
the combustion of fuel is complete.
1.2 State TWO functions of a steam condenser.

## QUESTION 2

Air in a cylinder has a pressure of 95 kPa and a temperature of $38^{\circ} \mathrm{C}$. The air is compressed according to the law $\mathrm{PV}^{1,24}=\mathrm{C}$, until the pressure is 650 kPa . The initial volume of air is $1,25 \mathrm{~m}^{3}$. Take the gas constant $R$ as $0,287 \mathrm{~kJ} / \mathrm{kg}$.

Calculate the following:
2.1 The mass of the air
2.2 The final volume of the air after compression
2.3 The final absolute temperature of the air after compression
2.4 The work done on the air during compression

## QUESTION 3

Feedwater enters an economiser at a temperature of $32,9^{\circ} \mathrm{C}$ in a steam boiler plant. The boiler has a pressure of 2500 kPa and uses 650 kg of coal per hour. The calorific value of the coal is $33 \mathrm{MJ} / \mathrm{kg}$. Feed water from the economiser enters the boiler at a temperature of $90{ }^{\circ} \mathrm{C}$. The boiler produces 5000 kg of wet steam per hour and the steam is 0,95 dry.

Calculate the following:
3.1 The quantity of heat absorbed by the feed water in the economiser
3.2 The thermal efficiency of the boiler
3.3 The equivalent evaporation (EE) from and at $100^{\circ} \mathrm{C}$

## QUESTION 4

The barometer and the manometer readings of a surface condenser are 755 mmHg and 650 mmHg respectively. The condenser receives 200 kg of steam per minute and is $10 \%$ wet. The inlet temperature of the cooling water is $15^{\circ} \mathrm{C}$ and the outlet temperature is $23{ }^{\circ} \mathrm{C}$. The condensate leaves the condenser at a temperature of $45,8^{\circ} \mathrm{C}$. Take the specific heat capacity of water as $4,187 \mathrm{~kJ} / \mathrm{kg} . \mathrm{K}$.

Calculate the following:
4.1 The absolute pressure
4.2 The mass of cooling water
4.3 The logarithmic temperature difference

## QUESTION 5

A liquid fuel $\mathrm{C}_{7} \mathrm{H}_{18}$ of 1 kg mass with $40 \%$ excess air is completely burned. The atomic masses of these elements are as follows:

Hydrogen $=1$
Oxygen $=16$
Carbon $=12$
Assume that air consists of $23 \%$ oxygen and $77 \%$ nitrogen by mass.
Calculate the following:
5.1 The theoretical mass of air required for complete combustion of 1 kg of the fuel
5.2 The actual mass of air required for the combustion of 1 kg of the fuel
5.3 The mass of each product of combustion
5.4 The percentage of each mass of combustion

## QUESTION 6

1 kg mass of air enters a single-cylinder, single-stage air compressor at a pressure of 103 kPa and a temperature of $22^{\circ} \mathrm{C}$. The air is then compressed according to the law $P V^{n}=C$ to a pressure of 900 kPa and a volume which is $\frac{1}{6}$ of the initial volume. Ignore the clearance volume. Take R for air as $0,287 \mathrm{~kJ} / \mathrm{kg} . \mathrm{K}$.

Calculate the following:
6.1 The compression index ( n )
6.2 The absolute temperature compression
6.3 The volume after compression
6.4 The work done per kg of air

## QUESTION 7

The velocity of a gas leaving the nozzle of a single-stage impulse turbine is $900 \mathrm{~m} / \mathrm{s}$ and the nozzle angle is $20^{\circ}$. The blade velocity is $300 \mathrm{~m} / \mathrm{s}$ and the blade velocity of coefficient is 0,7 . The mass flow rate of the gas is $1 \mathrm{~kg} / \mathrm{s}$. The inlet and the outlet angles of the moving blade are equal.
7.1 Construct a velocity diagram in the ANSWER BOOK (landscape) and enter ALL the values ( $\mathrm{m} / \mathrm{s}$ ) onto the diagram.

Use the scale $1 \mathrm{~mm}: 5 \mathrm{~m} / \mathrm{s}$
NOTE: NO marks will be given if the values are NOT indicated on the diagram and if the diagram is NOT constructed to scale.
7.2 Determine the following from the velocity diagram:
7.2.1 The inlet velocity of the moving blade
7.2.2 The outlet velocity of the moving blade
7.2.3 The axial thrust
7.2.4 $\quad$ The diagram efficiency

## POWER MACHINES N5

## FORMULA SHEET

1. $Q=W+\Delta U$
2. $\Delta U=m C_{v} \Delta T$
3. $Q=m C_{v} \Delta T$
4. $Q=m C_{p} \Delta T$
5. $Q=P_{1} V_{1} \ln \frac{V_{2}}{V_{1}}$
6. $\Delta S=m\left(C_{\nu} \ln \frac{T_{2}}{T_{1}}+R \ln \frac{V_{2}}{V_{1}}\right)$
7. $W=P_{1} \Delta V$
8. $W=P_{1} V_{1} \ln \frac{V_{2}}{V_{1}}$
9. $W=\frac{P_{1} V_{1}-P_{2} V_{2}}{n-1}$
10. $W=\frac{P_{1} V_{1}-P_{2} V_{2}}{\gamma-1}$
11. $R=C_{p}-C_{v}$
12. $\gamma=\frac{C_{p}}{C_{v}}$
13. $\frac{P_{1} V_{1}}{T_{1}}=\frac{P_{2} V_{2}}{T_{2}}$
14. $\quad P V=m R T$
15. $\quad P_{1} V_{1}=P_{2} V_{2}$
16. $\quad P_{1} V_{1}^{n}=P_{2} V_{2}^{n}$
17. $\frac{T_{2}}{T_{1}}=\left(\frac{P_{2}}{P_{1}}\right)^{\frac{n-1}{n}}=\left(\frac{V_{1}}{V_{2}}\right)^{n-1}$
18. $\quad \frac{T_{2}}{T_{1}}=\left(\frac{P_{2}}{P_{1}}\right)^{\frac{\gamma-1}{\gamma}}=\left(\frac{V_{1}}{V_{2}}\right)^{\gamma-1}$
19. $h=h_{f}+\chi h_{f g}$
20. $h=h_{g}+C_{p} \Delta T$
21. $h=h_{f}+h_{f g}=h_{g}$
22. 
23. 

$\chi=\frac{V_{m}}{V_{g}}$
24. $\chi=\frac{M}{M+m}$
25. $U=H-P V$
26. $g Z_{1}+U_{1}+P_{1} V_{1}+\frac{1}{2} C_{1}^{2}+Q=$ $g Z_{2}+U_{2}+P_{2} V_{2}+\frac{1}{2} C_{2}{ }^{2}+W$
27. $\eta=\frac{\dot{m}_{s}\left(h_{2}-h_{1}\right)}{\dot{m}_{f} C V}$
28. $E E=\frac{\dot{m}_{s}\left(h_{2}-h_{1}\right)}{\dot{m}_{f} 2257}$
29. $p=\left(B_{m} \pm M_{m}\right) \frac{101,325}{760}$
30. $m=\frac{100}{23}\left[C \frac{8}{3}+8 \mathrm{H}_{2}+\mathrm{S}-\mathrm{O}_{2}\right]$
31.

$$
\mathrm{C}_{x} \mathrm{H}_{y}+\left(x+\frac{y}{4}\right) \mathrm{O}_{2}=x \mathrm{CO}_{2}+\frac{y}{2} \mathrm{H}_{2} \mathrm{O}
$$

32. $H . C . V .=\left(C V_{C} . C\right)+C V_{H_{2}}\left(H_{2}-\frac{O_{2}}{8}\right)+\left(C V_{s} . S\right)$
33. 

$$
\text { L.C.V. }=\text { H.C.V. }-h_{f g}\left(9 H_{2}\right)
$$

34. $H . C . V .=\frac{\left(m_{w}+m_{e}\right) C_{p} \Delta T}{m_{f}}$
35. $\quad W=P_{1} V_{e}\left(\frac{n}{n-1}\right)\left[\left(\frac{P_{2}}{P_{1}}\right)^{\frac{n-1}{n}}-1\right]=m R T_{1}\left(\frac{n}{n-1}\right)\left[\left(\frac{P_{2}}{P_{1}}\right)^{\frac{n-1}{n}}-1\right]$
36. 

$\eta_{c}=\frac{V_{e}}{V_{s}} \cdot 100=1-\frac{V c}{V s}\left[\left(\frac{P_{2}}{P_{1}}\right)^{\frac{1}{n}}-1\right]=1+\alpha-\alpha\left(r_{p}\right)^{\frac{1}{n}}$
37. $\eta_{\alpha}=\frac{V_{\alpha}}{V_{s}} .100$
38.

$$
F_{c}=\dot{m}\left(C_{f e}-C_{f i}\right)
$$

39. 

$$
P=\dot{m} U\left[C_{w i}-\left(-C_{w e}\right)\right]
$$

40. $\quad \eta=\frac{2 U\left[C_{w i}-\left(-C_{w e}\right)\right]}{C_{a i}{ }^{2}} .100$
41. $\quad U=\pi D N$
42. $\dot{m} V=A C$
43. $(m+M) g=m \omega^{2} h$
44. $\quad V s=\frac{\pi}{4} D^{2} L$
45. $\quad \theta_{1}=t_{c}-t w i$
46. $\theta_{2}=t_{c}-t w o$
47. Log.temp.diff. $=\frac{\theta_{1}-\theta_{2}}{\ln \frac{\theta_{1}}{\theta_{2}}}$
48. $\quad$ Piso $=P_{1} V_{1} \ln \left(\frac{P_{2}}{P_{1}}\right)$
49. $\quad$ Pact $=\frac{n}{n-1} P_{1} V_{1}\left[\left(\frac{P_{2}}{P_{1}}\right)^{\frac{n-1}{n}}-1\right]$
50. $\quad$ Niso $=\frac{\text { Piso }}{\text { Pact }} \bullet 100$

